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(54) Title: COATING COMPOSITION FOR METAL SUBSTRATES (57) Abstract <p>The present invention relates to a composition for coating a metal substrate which is intended to be fabricated and overcoated, and wherein the binder comprises an aqueous silica sol or alkali metal silicate having a SiO₂/M₂O mole ratio of at least 6:1, where M represents total alkali metal and ammonium ions, and wherein the silica particles have an average size equal to or smaller than 10 nm. Very good results are obtained if the dried coating is post-treated with a film-strengthening solution.</p>		

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Coating composition for metal substrates

5 This invention relates to a coating composition that can be used for the coating of metal substrates, for example steel substrates. In particular, it relates to a coating composition for semi-finished steel products which are subsequently to be fabricated by heat-intensive processes and overcoated. Such semi-finished steel products are used in the shipbuilding industry and for other large-scale
10 structures such as oil production platforms and include steel plates, for example of thickness 6 to 75 mm, bars, girders and various steel sections used as stiffening members. The most important heat-intensive process is welding; substantially all such semi-finished steel products are welded. Other important heat-intensive processes are cutting, for example oxy-fuel cutting, plasma
15 cutting or laser cutting, and heat fairing in which the steel is bent into shape while being heated. These steel products are often exposed to the weather during storage before construction and during construction, and they are generally coated with a coating called a "shop primer" or "pre-construction coating" to avoid corrosion of the steel occurring before the steel construction,
20 e.g. ship, is given its full coating of anticorrosive paint, thereby avoiding the problem of having to coat over or remove steel corrosion products. In most big shipyards, the shop primer is applied as one of several treatments carried out on a production line in which the steel is for example preheated, shot- or grit-blasted to remove millscale and corrosion products, shop primed and passed
25 through a drying booth. The shop primer can alternatively be applied by a trade coater or steel supplier before the steel is delivered to the shipyard or other construction site.

Although the main purpose of the shop primer is to provide temporary corrosion
30 protection during construction, it is preferred by shipbuilders that the shop primer does not need to be removed but can remain on the steel during and after fabrication. Steel coated with shop primer thus needs to be weldable

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without removal of shop primer and to be overcoatable with the types of protective anti-corrosive coatings generally used on ships and other steel constructions, with good adhesion between the primer and the subsequently applied coating. The shop primed steel should preferably be weldable without significant detrimental effect on the quality of the weld or on the speed of the welding process and should be sufficiently resistant to heat that the shop primer retains its anticorrosive properties in areas heated during fairing or during welding of the opposite face of the steel.

- Commercially successful shop primers available today are solvent borne coatings based on prehydrolysed tetraethyl orthosilicate binders and zinc powder. Such coatings contain a large proportion of volatile organic solvent, typically about 650 grams per liter, to stabilize the paint binder and to enable the product to be applied as a thin film, typically about 20 microns thick. Release of volatile organic solvent can be damaging to the environment and is regulated by legislation in many countries. There is a need for a shop primer which releases no, or much less, volatile organic solvent. Examples of such coatings are described in US-A-4888056 and JP-A-7-70476.
- JP-A-6-200188 is concerned with shop primer coatings and mentions the possibility of using an aqueous alkali silicate salt type binder. Coatings comprising an aqueous alkali metal silicate and zinc powder are also proposed in GB-A-1226360, GB-A-1007481, GB-A-997094, US-A-4230496 and JP-A-55-106271. Alkali silicate binders for anticorrosive coatings are also mentioned in US-A-3522066, US-A-3620784, US-A-4162169 and US-A-4479824. We have found that primer coatings based on an aqueous alkali silicate binder containing zinc powder can give adequate corrosion protection and allow the steel surfaces they cover to be welded but give rise to problems when overcoated. The aqueous silicates contain a large quantity of alkali metal cations which are required to keep the silicate in aqueous solution and these ions are still present in the coating after the coating has dried. We have found that, if primer coatings

having these large quantities of alkali metal ions are overcoated with any conventional organic coating and then immersed in water, blistering (local delamination of the coating) occurs. We have performed tests that show that this problem can be reduced if the coating is weathered outside for some time
5 after application of the shop primer or washed prior to overcoating. However, these processes are not compatible with use in today's high productivity shipyards.

Aqueous silica sols having very low alkali metal ion content are available
10 commercially but coatings based on such sols have very poor (initial) film strength in terms of adhesion, cohesion, hardness, and resistance to abrasion and water. These poor physical properties of the coating make the coating susceptible to damage during handling or further processing. This brings the potential requirement for significant coating repair with major cost implications.
15 Suggested improvements to silica sol coatings are described in US-A-3320082, which adds a water-immiscible organic amine, GB-A-1541022, which adds a water-soluble acrylamide polymer and GB-A-1485169, which adds a quaternary ammonium or alkali metal silicate, but such coatings have not achieved physical properties similar to those of coatings based on alkali metal silicates. Coatings
20 based on silica sols show low levels of blistering when overcoated/immersed. Although the water soluble salt content and osmotic pressure is low, blistering can still occur as the coating exhibits little resistance to blister initiation/growth due to its poor physical properties. Further, these coatings exhibit a low abrasion resistance.

25

There is a need for a water-based shop primer of low alkali metal ion content which has improved adhesion to substrates and film strength in terms of the properties discussed above to resist blister initiation and growth.

Further, there is a need for a blister-free water-based shop primer showing
30 faster development of the physical properties of the coating after application of

the shop primer to enable the handling and further processing of the substrate without the risk of damaging the coating.

The coating composition according to the present invention provides a solution
5 to the above-mentioned problems/disadvantages. The composition for coating a metal substrate which is intended to be fabricated and overcoated according to the present invention comprises an aqueous silica sol or alkali metal silicate binder having a $\text{SiO}_2/\text{M}_2\text{O}$ mole ratio of at least 6:1, where M represents total alkali metal and ammonium ions, and wherein the silica particles have an
10 average size equal to or smaller than 10 nm.

The binder is most preferably based on an aqueous silica sol. Such sols are available from Akzo Nobel under the Registered Trademark "Bindzil" or from DuPont under the Registered Trademark "Ludox", although the literature
15 concerning them emphasizes that conventional grades of colloidal silica are not good film formers. Various grades of sol are available having various particle sizes of the colloidal silica and containing various stabilizers. The particle size of the colloidal silica can for example be in the range 3 to 100 nm. We found that coatings with good film properties can be obtained by using silica particles with
20 sizes towards the lower end of this range, viz. equal to or smaller than 10 nm, preferably in the range of 3 to 10 nm, more preferably in the range of 5 to 10 nm. To further improve the resistance to blister initiation and/or growth, the silica sol preferably has a $\text{SiO}_2/\text{M}_2\text{O}$ mole ratio of at least 25:1, more preferably at least 50:1 and may have a $\text{SiO}_2/\text{M}_2\text{O}$ mole ratio of 200:1 or more. Further, it
25 is possible to use a blend of two or more silica sols having a different $\text{SiO}_2/\text{M}_2\text{O}$ mole ratio, wherein the $\text{SiO}_2/\text{M}_2\text{O}$ mole ratio of the blend is at least 25:1. The sol can be stabilized by alkali, for example sodium, potassium, or lithium hydroxide or quaternary ammonium hydroxide, or by a water-soluble organic amine such as alkanolamine. The coating composition should preferably be substantially
30 free of any ammonium stabilized silica sol, since the presence of an ammonium stabilized sol could result in gelling of the composition, in particular when the

binder mainly consists of ammonium stabilized silica sol and the coating composition also contains zinc powder.

The silica sol can be blended with an alkali metal silicate, for example lithium silicate, sodium-lithium silicate or potassium silicate, or with ammonium silicate or a quaternary ammonium silicate. Other examples of suitable sol-silicate blends or mixtures can be found in US 4,902,442. The addition of an alkali metal or ammonium silicate may improve the initial film-forming properties of the silica sol, but the amount of alkali metal silicate should be low enough to keep the $\text{SiO}_2/\text{M}_2\text{O}$ mole ratio of the binder sol at least 6:1, preferably at least 8:1 and most preferably above 15:1.

The silica sol can alternatively or additionally contain a dissolved or dispersed organic resin. The organic resin is preferably a latex, for example a styrene butadiene copolymer latex, a styrene acrylic copolymer latex, a vinyl acetate ethylene copolymer latex, a polyvinyl butyral dispersion, a silicone/siloxane dispersion, or an acrylic based latex dispersion. Examples of suitable latex dispersions that can be used include XZ94770 and XZ94755 (both ex. Dow Chemicals), Airflex[®] 500, Airflex[®] EP3333 DEV, Airflex[®] CEF 52, and Flexcryl[®] SAF34 (all ex. Air Products), Primal[®] E-330DF and Primal[®] MV23 LO (both ex. Rohm and Haas), and Silres[®] MP42 E, Silres[®] M50E, and SLM 43164 (all ex. Wacker Chemicals). Water soluble polymers such as acrylamide polymers can be used but are less preferred. The organic resin is preferably used at up to 35% by weight, preferably 1-20% by weight, based on silica. Higher amounts may cause weld porosity during subsequent welding. It was found that the addition of an organic resin improves the adhesion/cohesion as measured in the cross hatch test.

The silica sol can alternatively contain a silane coupling agent which contains alkoxy silane groups and an organic moiety containing a functional group such as amino, epoxide or isocyanate group. The silane coupling agent is preferably

an aminosilane such as gamma-aminopropyl triethoxy silane or gamma-aminopropyl trimethoxy silane, or a partial hydrolysate thereof, although an epoxy silane such as gamma-glycidoxypopyl trimethoxy silane can also be used. The silane coupling agent is preferably present at up to 30% by weight, for example 1-20% by weight, based on silica.

The binder of the primer coating can alternatively comprise an aqueous solution of an alkali metal or ammonium silicate stabilized by a siliconate substituted by at least one anionic group of lower pKa than silicic acid such as a carboxylate or sulphonate group. Such a binder is preferably a solution having a $\text{SiO}_2/\text{M}_2\text{O}$ mole ratio in the range 8:1 to 30:1 and a pH of 7 to 10.5 prepared by lowering the pH of a solution of silicate and siliconate by cation exchange. For example the siliconate can be added at relatively low levels, for example at a molar ratio of 1:2 to 1:20, to a conventional 3.9:1 $\text{SiO}_2/\text{K}_2\text{O}$ alkali silicate. The solids may then be reduced to improve ease of processing and to further improve stability. At this stage the solution has a pH of 12-12.5. The solution is ion exchanged using a standard ion exchange resin. K^+ ions are replaced with H^+ reducing both the alkali content of the binder and the pH. Without the presence of the siliconate the silicate would gel on reducing the pH. Clear, stable solutions with pH's as low as 8 have been obtained. The resultant binder has an $\text{SiO}_2/\text{K}_2\text{O}$ mole ratio typically in the range 8-20:1 and can be concentrated if desired to increase the solids. The binder is a clear, stable solution and is stable in the presence of zinc but coatings based on these ion exchanged binders have relatively poor film strength, compared to coatings based on alkali silicate binders.

The primer coating preferably contains zinc powder, which is preferably of volume averaged mean particle size 2 to 12 microns and most preferably a product that is sold commercially as zinc dust having a mean particle size of 2 to 8 microns. The zinc powder protects the steel by a galvanic mechanism and may also form a protective layer of zinc corrosion products which enhances the

corrosion protection given by the coating. All or part of the zinc powder can be replaced by a zinc alloy. The amount of zinc powder and/or alloy in the coating is generally at least 10% and may be up to 90% by volume of the coating, on a dry film basis. The zinc powder and/or alloy can be substantially the whole of the pigmentation of the coating or can for example comprise up to 70%, for example 25 to 55%, by volume of the coating, on a dry film basis with the coating also containing an auxiliary corrosion inhibitor, which can for example be a molybdate, phosphate, tungstate or vanadate, as described in US-A-5246488 and/or a filler such as silica, calcined clay, alumina silicate, talc, barytes or mica. However, other pigments can be used in conjunction with zinc-based pigments. Examples of these other non-zinc pigments include conductive extenders such as di-iron phosphide (Ferrophos®), micaceous iron oxide etc. Use of these conductive non-zinc extender pigments may allow a reduction in the level of zinc while maintaining effective corrosion protection. To obtain optimum coating properties, extenders with a mean particle size below 3 μm , preferably below 2 μm , and small size pigments should be used.

The pigment volume concentration of the primer coating is preferably at least equal to the critical pigment volume concentration, as is usual for zinc silicate primers, for example 1.0 to 1.5 times the critical pigment volume concentration.

The pigment volume concentration (PVC) is the volume percent of pigment in the dry paint film. The critical pigment volume concentration (CPVC) is normally defined as the pigment volume concentration where there is just sufficient binder to provide a complete adsorbed layer of binder on the pigment surfaces and to fill all the interstices between the particles in a close-packed system. The critical pigment volume concentration can be determined by wetting out dry pigment with just sufficient linseed oil to form a coherent mass. This method yields a value known as the "oil absorption" from which the critical pigment volume concentration can be calculated. The method for determining oil absorption is described in British Standard 3483 (BS3483).

The solids content of the primer coating is generally at least 15% by volume and preferably in the range 20 to 40% by volume, more preferably in the range 25 to 35% by volume. The volume solids content is the theoretical value calculated on basis of all components that are present in the coating composition. The coating
5 preferably has a viscosity such that it can easily be applied by conventional coating applicators such as spray, particularly airless spray or high volume low pressure (HVLP) spray applicators, to give a coating of dry film thickness less than 40 microns, preferably between 12 and 25 to 30 microns.

Optionally, the coating composition may comprise further additives well known
10 to the skilled person, like, e.g., rheology control agents (organo clays, xanthanum gum, cellulose thickeners etc.), defoamers (in particular when latex modifiers are present), and pot-life extenders, such as chromates (for example sodium dichromate). If no pot-life extenders are present, the coating composition normally has a pot-life between 2-4 hours. Usually it is sufficient to
15 add only a small quantity (0.0125 - 0.025 wt% based on liquid paint) of sodium dichromate to obtain a pot life of greater than 24 hours. Higher levels normally lead to poor properties of the coating. Normally the coating system is provided as a two (or more) component system.

It is possible to prepare the coating composition just prior to application of the
20 coating, for example by adding and thorough mixing of all components of the coating composition shortly before application. Such process can also be referred to as on-line mixing of the components that are present in the coating composition. This process is in particular suited for coating compositions that have a limited pot-life.

25

In a preferred embodiment the coating composition according to the present invention is a water-based shop primer for the coating of steel substrates which are intended to be fabricated and overcoated, said composition having a solid content of 20 - 40 % by volume, comprising:

- 30 - an aqueous silica sol or alkali metal silicate binder having a $\text{SiO}_2/\text{M}_2\text{O}$ mole ratio of at least 6:1, where M represents total alkali metal and ammonium.

ions, and wherein the silica or silicate particles have an average size equal to or smaller than 10 nm,

- 10 - 90 % by volume of the coating on a dry film basis of zinc powder and/or a zinc alloy having a mean particle size in the range 2 to 12 μm ,
- 5 - 0 - 35 % by weight, based on silica or silicate, of an organic resin,
- 0 - 30 % by weight, based on silica or silicate, of a silane coupling agent,
- optionally non-zinc pigment(s) having a mean particle size below 3 μm , and
- optionally a pot-life extender.

10 It was found that the advantages of the coating composition according to the present invention, viz. improved adhesion to the substrate, film strength, and/or faster development of physical properties of the coating after application, can be illustrated by treating the coating with a film strengthening solution once the coating is dried to the extent that it is touch dry.

15 For the purpose of the present invention, a film strengthening solution is a solution that enhances the film strength of a primer coating and/or accelerates the development of the film strength with time.

The time to touch dry is generally about 10 to 15 minutes at ambient temperatures of 15 to 20°C or 3 to 4 minutes at 40°C for a 15-20 μm dry film
20 thickness (dft) coating. Drying time is also dependent on air flow and film thickness. At 35°C and 0.5 m/s air flow, the drying time for a 20 μm dry film thickness coating is approximately 2 minutes. This time can be further reduced by increasing the air temperature.

In general, the drying time can be reduced by either increasing the substrate
25 temperature, increasing the air temperature, using an air-flow, or by any combination thereof.

Application of a treatment solution before the primer is touch dry does not give film strengthening.

30 The solution which increases the film strength of the primer coating can in general be an aqueous solution of an inorganic salt or a solution of material

having reactive silicon-containing groups. The increase in film strength can be detected by a significant increase in hardness, abrasion resistance and usually adhesion. Hardness can be measured in pencil hardness test British Standard 3900, part E19 (1999) (hardness of pencil required to gouge the coating).
5 Abrasion resistance can be measured using a double rub test which automatically rubs the coating and can be carried out dry or wet with water. While a significant increase in either dry or wet abrasion resistance would be regarded as an increase in film strength of the primer coating, we have found that the treatment according to the invention generally increases both dry and
10 wet abrasion resistance. Adhesion can be measured by a cross-hatch test as described in British Standard 3900, part E6 (1992).

The amount of film strengthening solution applied to the primer coating is generally in the range 0.005-0.2, preferably 0.01 - 0.08 liters per square meter
15 of primer coated surface (L/m^2) for coatings applied at standard dry film thickness (15-20 μm). Such an amount of solution can conveniently be applied by spraying. Needless to say that the concentration or the volume of the post-treatment solution should be increased if the coating is over-applied, i.e. at a dry film thickness $> 20 \mu m$.

20

While we do not wish to be bound by any theory explaining the film strengthening, it appears that when the treatment solution is an aqueous inorganic salt solution, either silica dissolution and reprecipitation takes place or the salt acts as a reinforcing agent between the sol particles. When the
25 treatment solution contains reactive silica species, these can be deposited between the silica sol particles to improve bonding of the sol particles. We have found that the same strengthening materials, when added to the primer coating composition at or before application to the substrate, do not strengthen the primer coating film formed.

30

When the film strengthening solution is an aqueous solution of an inorganic salt, it generally has a concentration of at least 0.01M and preferably at least 0.03M. The concentration of the inorganic salt solution can be up to 0.5M or 1M or even higher. The inorganic salt can be the salt of a monovalent cation such as an alkali metal or ammonium salt, of a divalent cation such as zinc, magnesium, calcium, copper (II) or iron (II), of a trivalent cation such as aluminium or cerium (III), or of a tetravalent cations such as cerium (IV), and of a monovalent anion such as a halide, for example fluoride, chloride or bromide, or nitrate, or a polyvalent anion such as sulphate or phosphate. Mixtures of the above-mentioned salts can also be used. Examples of inorganic salt solutions which have been found effective are magnesium sulphate, zinc sulphate, potassium sulphate, aluminium sulphate, iron sulphate, cerium(IV)sulphate, copper sulphate, sodium chloride and potassium chloride, although chlorides may not be preferred because of their tendency to promote corrosion. The concentration of the inorganic salt solution in weight terms is preferably in the range 0.5-20 % by weight.

One example of a material having active silicon-containing groups is a silicate. The film strengthening solution can for example be an alkali metal silicate solution, for example potassium silicate or lithium silicate, or an ammonium silicate solution or can be an alkali metal silicate, for example an alkylsilicate solution. The preferred concentration of such a solution is in the range 0.5-20% by weight.

When the film strengthening solution is a solution of an inorganic salt or alkali metal silicate, the added material will increase the salt content of the zinc silicate primer coating. This will tend to increase the osmotic driving force when the coating is overcoated and thus the possibility of osmotic blistering when the coated substrate is immersed. The amount of inorganic salt or alkali metal silicate applied is preferably low enough so that the $\text{SiO}_2/\text{M}_2\text{O}$ mole ratio of the primer coating binder is kept above 6:1, preferably above 8:1 and most

preferably above 10:1. To achieve this, the amount of inorganic salt or alkali metal silicate applied in the film strengthening solution is preferably less than 10g/m^2 on a dry weight basis, most preferably less than 5g/m^2 , for a coating with a dry film thickness of 15 - 20 μm .

5 An alternative example of a material having reactive silicon-containing groups is an alkoxysilane or an acyloxysilane, for example acetoxysilane. This can for example be a tetraalkoxysilane (alkyl orthosilicate) such as tetraethoxysilane or tetraisopropoxysilane, or a trialkoxysilane such as methyl trimethoxysilane
10 (MTMS, ex Aldrich) or bistrimethoxysilane ethane. The alkoxysilane may contain additional functional groups, for example a trialkoxysilane can have the formula $\text{RSi}(\text{OR}^1)_3$ where each R^1 group is 1-3C alkyl and R is an alkyl or aryl group substituted by an amino, alkylamino, dialkylamino, amide, halogen, carbamate, epoxide, isocyanate, aziridine, sulphonate, carboxylate, phosphate
15 or hydroxyl group. Preferred examples are aminosilanes such as triethoxy silylpropylamine (Aminosilane A1100 ex. Witco), trimethoxy silylpropylamine (Aminosilane A1110 ex. Witco), trimethoxy silylpropylethylenediamine (Aminosilane A1120 ex. Witco), trimethoxy silylpropyldiethylenetriamine (Aminosilane A1130 ex. Witco) or bistrimethoxysilyl propylethylene diamine.
20 Further, the alkoxysilane can be a bis(trialkoxysilane), for example an alkylene or polydimethylsilane chain tipped with $-\text{SiOR}^1_3$ groups. The alkoxysilane can be at least partially hydrolyzed, for example a partially hydrolyzed tetraalkoxysilane or a hydrolyzed alkyl trialkoxy silane or aminoalkyl trialkoxy silane, can be used. The alkoxysilane is preferably applied from aqueous solution, although the
25 aqueous solution can contain a water-miscible organic solvent, for example an alcohol such as ethanol.

Further it was found that also orthosilicates are very effective property-enhancers in the post-treatment process. Aqueous solutions of tetramethylorthosilicate (TMOS) and tetraethylorthosilicate (TEOS) are effective
30 post-treatment agents. Better results are obtained if TMOS or TEOS are

hydrolysed at pH 1-2. At this pH the pot-life of the post treatment solution can even exceed 7 days.

The concentration of alkoxysilane or orthosilicates in the treatment solution is preferably in the range 1-25% by weight.

- 5 The use of alkoxysilanes and/or orthosilicates in the post-treatment solution is preferred since these compounds add virtually zero level of water soluble salts to the shop-primer.

The application of the treatment solution, and preferably also the drying of the
10 treated primer coating until the coating is again touch dry, can be carried out in an on-line process following primer coating of the steel and drying of the primer coating until it is touch dry. The amount of film strengthening solution applied is preferably 0.005-0.2 L/m² of primer coated surface, most preferably 0.08 L/m² or less if the coating is treated and dried on-line for a coating with a dry film
15 thickness of 15 - 20 µm. The drying time for this coating treated with such amount of film strengthening solution is generally about 5 to 10 minutes at 15-20°C or about 1.5 to 2 minutes at 40°C. The drying time can be further reduced by placing the primed substrate in an air flow.

In general, the drying time can be reduced by either increasing the substrate
20 temperature, increasing the air temperature, using an air-flow, or by any combination thereof.

The treatment solution can be applied by standard spray application equipment, for example airless spray or HVLP spray, or by a simple atomizer spray, simply by mounting a second spray gun further down the shop primer line from the
25 spray gun applying the primer. Alternatively, the solution can be applied using a mist coating application technique. The treatment solution can be applied to both sides of a substrate, for example to both sides of a steel plate for use in shipbuilding, whatever the orientation of the substrate; the volume of solution required to strengthen the film are such that the solution can be applied to the
30 underside of a plate without sagging or dripping. Other methods of application such as application by roller are possible but are not preferred. The treated

primer coating only needs to be allowed to dry on the substrate and does not need any subsequent washing or heating; once the treated primer is dry the coated product can be handled normally.

- 5 The treatment process in combination with the coating composition according to the invention increases the hardness, cohesion and abrasion resistance of the shop primer without introducing the disadvantage of blistering when overcoated. Further, the treatment process accelerates the development of these properties. This improves the resistance to damage during handling and fabrication on a
- 10 ship yard or steel mill. In addition to these benefits, the post-treated shop primed substrate shows the performance characteristics required in the shop primer market, viz. corrosion resistance of 6 months on outdoor exposure, excellent welding/cutting characteristics and overcoatability with a wide range of primer coatings with no blistering or pinholing.
- 15 For example when a zinc-filled silica sol coating is post-treated with a film strength enhancing solution, the dry abrasion resistance is increased at least fivefold and the wet abrasion resistance usually tenfold or more. The pencil hardness typically changes from 2B to H or harder. The $\text{SiO}_2/\text{M}_2\text{O}$ mole ratio of the primer coating may for example be reduced from 50-200 to 15-35 if the film
- 20 strengthening solution applied is an inorganic salt solution or alkali metal silicate solution, but at normal dry film thickness of 15-20 μm for shop primer coatings this is still above the level at which significant blistering occurs. The $\text{SiO}_2/\text{M}_2\text{O}$ mole ratio can be maintained at an even higher level if the film strengthening solution is an alkoxysilane solution. The treated primer coatings can be
- 25 overcoated with an amine-cured epoxy resin coating, or any other heavy duty coating such as a polyurethane, at a film thickness of 100 μm or 200 μm and after being allowed to cure for 7 days can be immersed in fresh or sea water for over 4 months at 40°C without blistering.

The invention will be elucidated with reference to the following examples. These are intended to illustrate the invention but are not to be construed as limiting in any manner the scope thereof.

5 The compounds used as starting material in the examples have the following origin:

Ludox® SM	a silica sol of concentration 30% by weight, particle size 7 nm, SiO ₂ /Na ₂ O mole ratio 50:1, ex DuPont
XZ 94770	a styrene/butadiene organic latex of 50 vol.% solids, ex. Dow Chemicals.
10 Huber 90C	a calcined aluminium silicate extender pigment of 0.7 µm mean particle size, ex JM Huber/Marlow Chemicals
Satintone® W	a calcined aluminium silicate extender pigment of 1.4 µm mean particle size, ex Lawrence Industries
15 Zinc dust	a 7 µm mean particle size metal powder, ex. Trident Alloys

Example 1

A primer coating of solids concentration 28% by volume was prepared from the following ingredients. The primer coating had a pigment volume concentration
20 of 74.6%, which is 1.3 times the critical pigment volume concentration ($\lambda = 1.3$).

Component	% by weight
Ludox® SM	27.3
water	15.8
Bentonite clay thixotrope	0.2
Zinc dust	48.2
Satintone	8.6

A primer was prepared by mixing the silica sol with water and thixotrope and the resulting binder was mixed with the pigments just before application to the steel, as is usual with zinc silicate coatings. The obtained primer coating was applied
25 to 15 cm x 10 cm steel panels at a dry film thickness of 15-20 µm. The primer

was allowed to dry at 30°C, 30% RH in an air-flow of 0.5 m/s and when touch dry the panels were sprayed on-line with 0.3 g of various solutions.

The treated panels were allowed to dry at 25°C, 75% RH and were tested for the physical properties 10 minutes and 1 hour after application of the various solutions.

The treated panels were tested for pencil hardness (British Standard 3900, part E19 = BS 3900:E19), cross-hatch adhesion (British Standard 3900, part E6 = BS 3900:E6) on a scale rated from 0 (no adhesion) to 5 (100% adhesion) and abrasion resistance (double rub test). In the double rub test the treated surface is wetted with a couple of drops of water (if doing wet double rubs) then rubbed with a cotton wool swab using light pressure. One pass to and fro is a double rub. The results are expressed as the number of double rubs till removal of the coating. If the coating survives 100 double rubs the final dry film thickness (dft) is compared to the initial value. If the dry film thickness is reduced by more than 25% the result is expressed as >100. If the dry film thickness is reduced by less than 25% the result is expressed as >>100. The results of the tests are shown in Table 1A and 1B.

Table 1A

Example No.	additive and concentration	Mechanical properties 10 minutes after treatment		
		Double rubs Dry/Wet	Pencil Hardness	Cross-hatch Adhesion
1a	5% Zinc sulphate	>>100/27-45	3-4H	3
1b	2% Aluminium sulphate	>>100/50-100	2H	1
1c	5% TMOS (pH 2)	>>100/45-70	H	1
1d	5% Amino silane A1120	>>100/19-31	HB	0

Table 1B

Example No.	additive and concentration	Mechanical properties 1 hour after treatment		
		Double rubs Dry/Wet	Pencil Hardness	Cross-hatch Adhesion
1a	5% Zinc sulphate	>>100/>>100	5H	4-5
1b	2% Aluminium sulphate	>>100/>>100	4-5H	0-4
1c	5% TMOS (pH 2)	>>100/>>100	3H	1-2
1d	5% Amino silane A1120	>>100/30-50	H	0

Example 2

- 5 A primer coating of solids concentration 28% by volume was prepared from the following ingredients. The primer coating had a pigment volume concentration of 74.6%, which is 1.3 times the critical pigment volume concentration ($\lambda = 1.3$).

Component	% by weight
Ludox [®] SM	22.0
XZ 94770	1.5
water	19.1
Bentonite clay thixotrope	0.2
Zinc dust	48.6
Satintone	8.6

- 10 A primer was prepared by mixing the silica sol with water and thixotrope and the resulting binder was mixed with the pigments just before application to the steel, as is usual with zinc silicate coatings. The obtained primer coating was applied to 15 cm x 10 cm steel panels at a dry film thickness of 15-20 μm . The primer was allowed to dry at 30°C, 30% RH in an air-flow of 0.5 m/s and when touch dry the panels were sprayed on-line with 0.3 g of various solutions.
- 15 The treated coating was allowed to dry at 25°C, 75% RH and was tested for physical properties 10 minutes and 1 hour after application of the various solutions. The results of the tests are shown in Table 2A and 2B.

Table 2A

Example No.	additive and concentration	Mechanical properties 10 minutes after treatment		
		Double rubs Dry/Wet	Pencil Hardness	Cross-hatch Adhesion
2a	5% Zinc sulphate	>>100/60-100	6H	4-5
2b	2% Aluminium sulphate	>>100/35-70	5H	3-4
2c	5% TMOS (pH 2)	>>100/19-31	H	3
2d	5% Amino silane A1120	>>100/14-22	2H	2

Table 2B

Example No.	additive and concentration	Mechanical properties 1 hour after treatment		
		Double rubs Dry/Wet	Pencil Hardness	Cross-hatch Adhesion
2a	5% Zinc sulphate	>>100/>>100	6H	5
2b	2% Aluminium sulphate	>>100/>>100	6H	3-4
2c	5% TMOS (pH 2)	>>100/35-50	4H	3-4
2d	5% Amino silane A1120	>>100/35-40	3H	2-3

5

Example 3

A primer coating of solids concentration 28% by volume was prepared from the following ingredients. The primer coating had a pigment volume concentration of 68.4%, which is 1.2 times the critical pigment volume concentration ($\lambda = 1.2$).

Component	% by weight
Ludox® SM	27.6
XZ 94770	1.8
water	15.1
Bentonite clay thixotrope	0.2
Zinc dust	48.9
Satintone	6.4

A primer was prepared by mixing the silica sol with water and thixotrope and the resulting binder was mixed with the pigments just before application to the steel, as is usual with zinc silicate coatings. The obtained primer coating was applied to 15 cm x 10 cm steel panels at a dry film thickness of 15-20 μm . The primer was allowed to dry at 30°C, 30% RH in an air-flow of 0.5 m/s and when touch dry the panels were sprayed on-line with 0.3 g of various solutions.

The treated coating was allowed to dry at 25°C, 75% RH and was tested for physical properties 10 minutes and 1 hour after application of the various solutions. The results of the tests are shown in Table 3A and 3B.

Table 3A

Example No.	additive and concentration	Mechanical properties 10 minutes after treatment		
		Double rubs Dry/Wet	Pencil Hardness	Cross-hatch Adhesion
3a	5% Zinc sulphate	>>100/>>100	6H	4-5
3b	2% Aluminium sulphate	>>100/>>100	2-5H	1-2
3c	5% TMOS (pH 2)	>>100/60-100	3H	3-4
3d	5% Amino silane A1120	>>100/35-60	H-2H	1-2

Table 3B

Example No.	additive and concentration	Mechanical properties 1 hour after treatment		
		Double rubs Dry/Wet	Pencil Hardness	Cross-hatch Adhesion
3a	5% Zinc sulphate	>>100/>>100	6H	4-5
3b	2% Aluminium sulphate	>>100/>>100	2-5H	2-3
3c	5% TMOS (pH 2)	>>100/>>100	3H	3-4
3d	5% Amino silane A1120	>>100/>100	H-2H	1-2

Example 4

A primer coating of solids concentration 28% by volume was prepared from the following ingredients. The primer coating had a pigment volume concentration of 71.5%, which is 1.4 times the critical pigment volume concentration ($\lambda = 1.4$).

Component	% by weight
Ludox [®] SM	24.7
XZ 94770	1.6
water	17.2
Bentonite clay thixotrope	0.2
Zinc dust	48.7
Huber 90C	7.6

- 5 A primer was prepared by mixing the silica sol with water and thixotrope and the resulting binder was mixed with the pigments just before application to the steel, as is usual with zinc silicate coatings. The obtained primer coating was applied to 15 cm x 10 cm steel panels at a dry film thickness of 15-20 μm . The primer was allowed to dry at 30°C, 30% RH in an air-flow of 0.5 m/s and when touch
- 10 dry the panels were sprayed on-line with 0.3 g of various solutions.
- The treated coating was allowed to dry at 25°C, 75% RH and was tested for physical properties 10 minutes and 1 hour after application of the various solutions. The results of the tests are shown in Table 4A and 4B.

15

Table 4A

Example No.	additive and concentration	Mechanical properties 10 minutes after treatment		
		Double rubs Dry/Wet	Pencil Hardness	Cross-hatch Adhesion
4a	5% Zinc sulphate	>>100/>>100	3H	5
4b	2% Aluminium sulphate	>>100/>>100	2-3H	4-5
4c	5% TMOS (pH 2)	>>100/>>100	4-6H	3-4
4d	5% Amino silane A1120	>>100/>>100	2-4H	4

Table 4B

Example No.	additive and concentration	Mechanical properties 1 hour after treatment		
		Double rubs Dry/Wet	Pencil Hardness	Cross-hatch Adhesion
4a	5% Zinc sulphate	>>100/>>100	4-5H	5
4b	2% Aluminium sulphate	>>100/>>100	6H	4-5
4c	5% TMOS (pH 2)	>>100/>>100	3H	3-4
4d	5% Amino silane A1120	>>100/>100	3H	3-4

Example 5

A primer coating of solids concentration 28% by volume was prepared from the following ingredients. The primer coating had a pigment volume concentration of 76.0%, which is 1.4 times the critical pigment volume concentration ($\lambda = 1.4$).

Component	% by weight
Ludox® SM	20.7
XZ 94770	1.3
water	19.9
Bentonite clay thixotrope	0.2
Zinc dust	48.3
Huber 90C	6.1
'Molywhite' molybdate pigment	3.5

A primer was prepared by mixing the silica sol with water and thixotrope and the resulting binder was mixed with the pigments just before application to the steel, as is usual with zinc silicate coatings. The obtained primer coating was applied to 15 cm x 10 cm steel panels at a dry film thickness of 15-20 μm . The primer was allowed to dry at 30°C, 30% RH in an air-flow of 0.5 m/s and when touch dry the panels were sprayed on-line with 0.3 g of various solutions.

The treated coating was allowed to dry at 25°C, 75% RH and was tested for physical properties 10 minutes and 1 hour after application of the various solutions. The results of the tests are shown in Table 5A and 5B.

Table 5A

Example No.	additive and concentration	Mechanical properties 10 minutes after treatment		
		Double rubs Dry/Wet	Pencil Hardness	Cross-hatch Adhesion
5a	5% Zinc sulphate	-/30-70	4H	3-4
5b	2% Aluminium sulphate	-/30-60	2-3H	3
5c	5% TMOS (pH 2)	-/60-75	2H	2
5d	5% Amino silane A1120	-/30-60	2H	1

Table 5B

Example No.	additive and concentration	Mechanical properties 1 hour after treatment		
		Double rubs Dry/Wet	Pencil Hardness	Cross-hatch Adhesion
5a	5% Zinc sulphate	-/>>100	6H	5
5b	2% Aluminium sulphate	-/>>100	3H	4-5
5c	5% TMOS (pH 2)	-/>>100	5H	4
5d	5% Amino silane A1120	-/>>100	4H	3

5

Example 6

A primer coating of solids concentration 25% by volume was prepared from the following ingredients. The primer coating had a pigment volume concentration of 74.6%, which is 1.3 times the critical pigment volume concentration ($\lambda = 1.3$).

Component	% by weight
Ludox® SM	25.8
water	20.4
Bentonite clay thixotrope	0.2
Zinc dust	45.5
Satintone	8.1

To evaluate the effect on coating properties of the addition of a pot-life extender, based on the above primer coating, shop primers were prepared having a different level of sodium dichromate. The primer coatings were applied to 15 cm x 10 cm steel panels at a dry film thickness of 15-20 μm and when touch dry post-treated with a 5% by weight ZnSO_4 solution. The panels were then stored at 20°C, 70% RH. The physical properties were measured 1 hour after application of the ZnSO_4 solution. The results of the tests are shown in Table 6.

The compositions comprising a pot-life extender all had a pot-life of greater than 24 hours. A comparable composition that is free of any pot-life extender in general has a pot-life between 2 to 4 hours.

Table 6

Example No.	$\text{Na}_2\text{Cr}_2\text{O}_7 \cdot 2\text{H}_2\text{O}$ added (% by weight in the wet paint)	Mechanical properties 1 hour after treatment		
		Double rubs Dry/Wet	Pencil Hardness	Cross-hatch Adhesion
6a	0.0125	>>100/45	3/4H	4-5
6b	0.025	90/18	3H	4
6c	0.125	40/10	< B	1-2
6d	0.25	8/3	< 3B	1-2

Example 7

A zinc free primer coating of solids concentration 28% by volume was prepared from the following ingredients. The primer coating had a pigment volume concentration of 68.0%, which is 1.3 times the critical pigment volume concentration ($\lambda = 1.3$).

Component	% by weight
Ludox [®] SM	42.5
water	13.5
Bentonite clay thixotrope	0.2
Barytes (FB11)	33.9
Satintone	9.9

The primer coating was applied to 15 cm x 10 cm steel panels at a dry film thickness of 15-20 μ m and when touch dry post-treated with different solutions.

- 5 The panels were then stored at 20°C, 70% RH. The physical properties were measured 10 minutes and 1 day after application of the solutions. The results of the tests are shown in Tables 7A and 7B.

Table 7A

Example No.	Treatment solution	Mechanical properties 10 minutes after treatment		
		Double rubs Dry/Wet	Pencil Hardness	Cross-hatch Adhesion
7a	5% ZnSO ₄	>>100/40-70	H	0
7b	5% TEOS (pH=2)	>>100/>>100	4-5H	0

10

Table 7B

Example No.	Treatment solution	Mechanical properties 1 day after treatment		
		Double rubs Dry/Wet	Pencil Hardness	Cross-hatch Adhesion
7a	5% ZnSO ₄	>>100/>>100	5-7H	0
7b	5% TEOS (pH=2)	>>100/>>100	6H	0

Example 8

A primer coatings of solids concentration 28% by volume were prepared analogous to the formulation of Example 1 using silica sols with different mean particle size, viz. Ludox[®] SM (mean particle size 7 nm), Ludox[®] HS30 (mean particle size 12 nm) and Bindzil 40/170[®] (mean particle size 20 nm).

The primer coatings were applied to 15 cm x 10 cm steel panels at a dry film thickness of 15-20 μ m and when touch dry post-treated with different solutions.

The panels were then stored at 30°C, 30% RH. The physical properties were measured 10 minutes and 1 hour after application of the solutions. The results of the tests are shown in Tables 8A and 8B.

Table 8A

Example No.	Silica type	Treatment solution	Mechanical properties 10 minutes after treatment		
			Double rubs Dry/Wet	Pencil Hardness	Cross-hatch Adhesion
8a	Ludox SM	5% ZnSO ₄	>>100/25	2-4H	1
8b	Ludox SM	5% TEOS (pH=2)	>>100/>100	3-4H	1
8c*	Ludox HS30	5% ZnSO ₄	50-60/8	3H	2
8d*	Ludox HS30	5% TEOS (pH=2)	50-60/8-12	2H	2
8e*	Bindzil	5% ZnSO ₄	12/3-4	2-3H	2
8f*	Bindzil	5% TEOS (pH=2)	20/6	3H	1

*) Comparative example

Table 8B

Example No.	Silica type	Treatment solution	Mechanical properties 1 hour after treatment		
			Double rubs - Dry/Wet	Pencil Hardness	Cross-hatch Adhesion
8a	Ludox SM	5% ZnSO ₄	>>100/>>100	2-4H	1
8b	Ludox SM	5% TEOS (pH=2)	>>100/>>100	5H	2
8c*	Ludox HS30	5% ZnSO ₄	-/5-12	5H	-
8d*	Ludox HS30	5% TEOS (pH=2)	-/12	3-4H	-
8e*	Bindzil	5% ZnSO ₄	-/4	4H	-
8f*	Bindzil	5% TEOS (pH=2)	-/7-8	4H	-

*) Comparative example

Example 9

- 5 A primer coatings of solids concentration 28% by volume were prepared analogous to the formulation of Example 2 using silica sols with different mean particle size, viz. Ludox[®] SM (mean particle size 7 nm), Ludox[®] HS30 (mean particle size 12 nm) and Bindzil 40/170[®] (mean particle size 20 nm).
- 10 The primer coatings were applied to 15 cm x 10 cm steel panels at a dry film thickness of 15-20 µm and when touch dry post-treated with different solutions. The panels were then stored at 20°C, 50% RH (average conditions). The physical properties were measured 10 minutes and 1 hour after application of the solutions. The results of the tests are shown in Tables 9A and 9B.

Table 9A

Example No.	Silica type	Treatment solution	Mechanical properties 10 minutes after treatment		
			Double rubs Dry/Wet	Pencil Hardness	Cross-hatch Adhesion
9a	Ludox SM	5% ZnSO ₄	>>100/60-100	6H	4-5
9b	Ludox SM	5% TEOS (pH=2)	>>100/>>100	4H	3
9c*	Ludox HS30	5% ZnSO ₄	>100/5-7	5H	PCS ¹
9d*	Ludox HS30	5% TEOS (pH=2)	>100/7-10	4-5H	PCS ¹
9e*	Bindzil	5% ZnSO ₄	9/3-4	3-4H	PCS ¹
9f*	Bindzil	5% TEOS (pH=2)	50/5	3-4H	PCS ¹

*) Comparative example
 1) Poor cohesive strength

5

Table 9B

Example No.	Silica type	Treatment solution	Mechanical properties 1 hour after treatment		
			Double rubs Dry/Wet	Pencil Hardness	Cross-hatch Adhesion
9a	Ludox SM	5% ZnSO ₄	>>100/>>100	6H	5
9b	Ludox SM	5% TEOS (pH=2)	>>100/>>100	4-5H	3-5
9c*	Ludox HS30	5% ZnSO ₄	>100/10	5H	PCS ¹
9d*	Ludox HS30	5% TEOS (pH=2)	>100/15	5H	PCS ¹
9e*	Bindzil	5% ZnSO ₄	14/7	5H	PCS ¹
9f*	Bindzil	5% TEOS (pH=2)	90/10	5H	PCS ¹

*) Comparative example
 1) Poor cohesive strength

Claims

1. A composition for coating a metal substrate which is intended to be fabricated and overcoated, said composition comprising a silica or silicate binder, characterized in that the binder comprises an aqueous silica sol or alkali metal silicate having a $\text{SiO}_2/\text{M}_2\text{O}$ mole ratio of at least 6:1, where M represents total alkali metal and ammonium ions, and wherein the silica or silicate particles have an average size equal to or smaller than 10 nm.
2. A coating composition according to claim 1, characterized in that the binder is a silica sol of $\text{SiO}_2/\text{M}_2\text{O}$ mole ratio at least 25:1.
3. A coating composition according to claim 1 or claim 2, characterized in that the binder comprises an aqueous solution of an alkali metal or ammonium silicate stabilized by a siliconate substituted by at least one anionic group of lower pK_a than silicic acid, having a pH of 7 to 10.5 prepared by lowering the pH of a solution of silicate and siliconate by ion exchange.
4. A coating composition according to any of claims 1 to 3, characterized in that the coating composition further comprises zinc powder and/or a zinc alloy.
5. A coating composition according to any of claims 1 to 4, characterized in that the silica particles have an average size in the range 3 nm to 10 nm.
6. A coating composition according to any of claims 1 to 5, characterized in that the binder further comprises a silane coupling agent.
7. A coating composition according to any of claims 1 to 6, characterized in that the binder further comprises an organic resin.

8. A coating composition according to any of claims 1 to 7, characterized in that it is a water-based shop primer.
- 5 9. Water-based shop primer for the coating of steel substrates which are intended to be fabricated and overcoated, said composition having a solid content of 20 - 40 % by volume, comprising:
- an aqueous silica sol or alkali metal silicate binder having a $\text{SiO}_2/\text{M}_2\text{O}$ mole ratio of at least 6:1, where M represents total alkali metal and ammonium ions, and wherein the silica or silicate particles have an
10 average size equal to or smaller than 10 nm,
 - 10 - 90 % by volume of the coating on a dry film basis of zinc powder and/or a zinc alloy having a mean particle size in the range 2 to 12 μm ,
 - 0 - 35 % by weight, based on silica or silicate, of an organic resin,
 - 15 - 0 - 30 % by weight, based on silica or silicate, of a silane coupling agent,
 - optionally non-zinc pigment(s) having a mean particle size below 3 μm , and
 - optionally a pot-life extender.

INTERNATIONAL SEARCH REPORT

International Application No
PCT/EP 00/02473

A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 C09D1/02

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C09D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
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A	US 4 277 284 A (GINSBERG THOMAS ET AL) 7 July 1981 (1981-07-07) column 2, line 8 -column 4, line 25	1,4
A	WO 98 58028 A (AMERON INT CORP ;MONTES ERNESTO (US)) 23 December 1998 (1998-12-23) page 1, line 9 page 3, paragraphs 25-30 page 4, line 30 -page 5, line 10 page 7, line 18-31 page 8, line 10-16	1
	— —/—	

☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

* Special categories of cited documents :

- "A" document defining the general state of the art which is not considered to be of particular relevance
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INTERNATIONAL SEARCH REPORT

International Application No
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C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
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information on patent family members

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